

## Course Information Sheet

<b>University:</b> University of Presov	
<b>Faculty:</b> Faculty of Humanities and Natural Sciences	
<b>Code:</b> 2EKO/ANACH/22	<b>Title of Course:</b> Analytical chemistry
<b>Form of Study:</b> lectures 1 hour per week, seminars and laboratory course 2 hours per week	
<b>Number of contact hours:</b>	
<b>per week:</b> 1+2 <b>per level/semester:</b> lectures 10 hours, seminars and laboratory course 20 hours, self-study 90 hours	
<b>Method of study:</b> full-time study	
<b>Number of credits:</b> 4	
<b>Semester:</b> 1.	
<b>Degree/Level:</b> 2.	
<b>Prerequisites:</b> -	
<b>Grading Policy (Assessment/Evaluation):</b>	
Written exam 70 %.	
Continuous semestral examination during seminars 30 %.	
<b>Success criteria (percentage):</b>	
a) A - 100.00 - 90.00%	
b) B - 89.99 - 80.00%	
c) C - 79.99 - 70.00%	
d) D - 69.99 - 60.00%	
e) E - 59.99 - 50.00%	
f) FX - 49.99 and less%.	
<b>Aims and Objectives:</b>	
After the completion of this course, students will:	
- be familiar with the methods of chemical analysis;	
- be able to perform basic calculations in analytical chemistry;	
- know to determine results of basic analytical measurements;	
- be able to perform basic laboratory experiments in analytical chemistry.	
<b>Syllabus/Indicative Content:</b>	
1. Analytical chemistry - general terminology, analytical reactions, preparation of solutions.	
2. Qualitative chemical analysis – common, selective and specific reagents for cations and anions.	
3. Quantitative chemical analysis – volumetric and gravimetric analysis.	
4. Volumetric titrations. Classification of titration methods.	
5. Acid-base titrations.	
6. Precipitation titrations.	
7. Complexometric titrations.	
8. Redox titrations.	
9. Electroanalytical methods - potentiometry, electrogravimetry, and coulometry.	
10. Optical methods in analytical chemistry.	
11. Spectral methods – FTIR, UV-VIS, NMR, AAS.	
12. Non-spectral optical methods – refractometry, polarimetry, RTG diffraction analysis.	
13. Purification and separation techniques – chromatography.	
14. Laboratory experiments in analytical chemistry.	
<b>Self-study:</b>	
1. Preparation to laboratory course (protocols).	

**2. Calculations in analytical chemistry.**

**Suggested readings:**

1. J. Garaj a kol.: *Analytická chémia*, SNTL/ALFA, Bratislava 1987
2. Majer L. a kol. *Analytická chémia*, 1989
3. Gondova T. a kol. *Analytická chémia*, Košice, 1998
4. A. Skoog, F.J. West, F.J.Holler: *Analytical Chemistry. An Introduction*, Saunders College, 6th ed., Philadelphia, 1994.
5. R.Keller (Ed.): *Analytical chemistry*, Wiley-VCH, Weinheim 1998.
6. Dean's *Analytical Chemistry Handbook*. Pradyot Patnaik, McGraw-Hill Education - Europe, 2004, 1280 p.
7. *Handbook of Green Analytical Chemistry*, Eds: Miguel de la Guardia Salvador Garrigues, 2012, ISBN: 9780470972014, John Wiley a Sons, Ltd, 566 p.

**Language of Instruction:** slovak

**Other course information:** available in winter semester

**Grading history**

83 students

A	B	C	D	E	FX
10%	12%	27%	22%	18%	12%

**Lecturer/Instructor:**

Doc. Ruslan Mariychuk, PhD. - lectures

RNDr. Romana Smolková, PhD. – seminars

RNDr. Adriana Eliašová, PhD. – semináre/cvičenia

RNDr. Daniela Gruľová, PhD. – semináre/cvičenia

**Last update:** 9. May 2022

**Approved by:** uvádzza sa meno a priezvisko zamestnanca vyskej školy (štandardne garant študijného programu), ktorý zmenu schválil